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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.	
10/578,617	05/08/2006	Kazunari Kobayashi	290768US2PCT	3715	
22850 7590 03/18/2009 OBLON, SPIVAK, MCCLELLAND MAIER & NEUSTADT, P.C. 1940 DUKE STREET ALEXANDRIA, VA 22314			EXAMINER		
			DAVIS, PATRICIA A		
ALEAANDRIA, VA 22314			ART UNIT	PAPER NUMBER	
			4111		
			NOTIFICATION DATE	DELIVERY MODE	
			03/18/2009	ELECTRONIC	

## Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail address(es):

patentdocket@oblon.com oblonpat@oblon.com jgardner@oblon.com

Office Action Summary		Applicat	Application No.		Applicant(s)		
		10/578,6	317	KOBAYASHI ET AL.			
		Examine	er	Art Unit			
		PATRIC	A DAVIS	4111			
Period fo	The MAILING DATE of this commun	nication appears on th	ne cover sheet wit	h the correspondence ac	ddress		
A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.  - Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.  - If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.  - Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).							
Status							
	Responsive to communication(s) file	ed on 08 May 2006					
2a)□	•	2b)⊠ This action is	non-final				
3)□		<i>′</i> —		ers prosecution as to the	e merits is		
٠,١	Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under <i>Ex parte Quayle</i> , 1935 C.D. 11, 453 O.G. 213.						
Dispositi	on of Claims						
4)⊠	Claim(s) <u>1-10</u> is/are pending in the	application.					
•	4a) Of the above claim(s) is/are withdrawn from consideration.						
	Claim(s) is/are allowed.						
· · · · · · · · · · · · · · · · · · ·	Claim(s) <u>1-10</u> is/are rejected.						
·	Claim(s) is/are objected to.						
•	Claim(s) are subject to restri	ction and/or election	requirement.				
Applicati	on Papers						
	The specification is objected to by the	ne Examiner					
,—	The drawing(s) filed on is/are		o)□ objected to b	ov the Examiner.			
٠٠/	<del></del>	·— ·	·— •				
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).  Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).							
11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.							
Priority under 35 U.S.C. § 119							
12)⊠ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).							
(۵	a) ☑ All b) ☐ Some * c) ☐ None of:						
	<ul> <li>1. ☐ Certified copies of the priority documents have been received.</li> <li>2. ☐ Certified copies of the priority documents have been received in Application No.</li> </ul>						
	3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).						
* See the attached detailed Office action for a list of the certified copies not received.							
222 m.s attached actained chief actain for a not of the continue copies not received.							
Attachmen	` '		4) 🖂 استمار ۲۰۰۰	Immorty (DTO 440)			
1) Notice of References Cited (PTO-892)  4) Interview Summary (PTO-413)  2) Notice of Draftsperson's Patent Drawing Review (PTO-948)  Paper No(s)/Mail Date							
3) Information Disclosure Statement(s) (PTO/SB/08)							
Paper No(s)/Mail Date <u>See Continuation Sheet</u> . 6)  Other:							

 $Continuation \ of \ Attachment(s)\ 3).\ Information \ Disclosure \ Statement(s)\ (PTO/SB/08),\ Paper\ No(s)/Mail\ Date : 5/8/2006; 8/8/2006; 1/9/2007; 7/18/2007.$ 

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## **DETAILED ACTION**

## Claim Rejections - 35 USC § 102

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless –

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

Claims 1-10 are rejected under 35 U.S.C. 102(b) as being unpatentable by Hikata et al. (JP 7-94193) (hereinafter "Hikata").

Regarding claim 1, Hikata teaches an active material for a battery anode, the material is made of zinc and virtually contains no lead (see paragraphs 0001 and 0010).

Although, Hikata does not specifically teach that the disclosed material exhibits the recited change in weight due to corrosion upon exposure to the electrolyte solution as claimed. However, regarding composition claims, if the composition is the same, it must have the same properties (see MPEP § 2112.01, II.). Consequently, as Hikata teaches the same material composition, it is inherently anticipated that the active material for the battery anode would exhibit the same properties as recited in the claim.

Regarding claim 2, Hikata teaches the active material having a concentration of 99.99% or more of zinc metal (see paragraph 0010).

Regarding claim 3, Hikata teaches the active material consists of zinc for the major substance with 0.01-0.07 percent mass of bismuth (1000 ppm or 0.1%) (see par. 0010 and Table 2, line 74).

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Regarding claim 4, Hikata teaches the active material consist of zinc for the major substance with addition and compound of 0.01-0.007 percent by mass of bismuth (1000 ppm or 0.1%), 0.0003-0.03 percent by mass of magnesium (100 ppm or 0.01%), and 0.001-0.05 percent by mass selected from zirconium, strontium, barium, indium, and aluminum (100 ppm or 0.01% strontium) (see par. 0010 and Table 2, line 74).

Regarding claim 5, Hikata teaches an active material for a battery anode, the material is made of zinc and virtually contains no lead (see pars. 0001 and 0010).

Although, Hikata does not specifically teach that the disclosed material exhibits the recited change in weight due to corrosion upon exposure to the electrolyte solution as claimed. However, regarding composition claims, if the composition is the same, it must have the same properties (see MPEP § 2112.01, II.). Consequently, as Hikata teaches the same material composition, it is inherently anticipated that the active material for the battery anode would exhibit the same properties as recited in the claim.

Regarding claim 6, Hikata teaches the active material having a purity of 99.99% by mass or more of zinc metal in major substance (see pars. 0001 and 0010).

Regarding claim 7, Hikata teaches the active material consists of zinc for the major substance with 0.01-0.07 percent mass of bismuth (1000 ppm or 0.1%) (see pars. 0001 and 0010 and Table 2, line 74).

Regarding claim 8, Hikata teaches the active material consist of zinc for the major substance with addition and compound of 0.01-0.007 percent by mass of bismuth (1000 ppm or 0.1%), 0.0003-0.03 percent by mass of magnesium (100 ppm or 0.01%), and 0.001-0.05 percent by mass selected from zirconium, strontium, barium, indium,

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and aluminum (100 ppm or 0.01% strontium) (see pars. 0001 and 0010 and Table 2, line 74).

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Regarding claim 9, Hikata teaches a method of manufacturing a manganese dry battery with use of an anode zinc plate which is processed from an anode active material sheet in a temperature in a range of 120-210 degree Centigrade (180-220 degree Centigrade) where the material contains zinc and the addition of bismuth (see pars. 0006 and 0013; Table 2, line 74; and claim 1).

Regarding claim 10, Hikata teaches a method of manufacturing a manganese dry battery with use of an anode zinc plate which is processed in a temperature ranging from 100-250 degrees Centigrade (180-200 degrees Centigrade) from an anode active material sheet which material contains zinc for major substance with addition and compound of 0.01-0.7 percent by mass of bismuth, 0.01-0.007 percent by mass of bismuth (1000 ppm or 0.1%), 0.0003-0.03 percent by mass of magnesium (100 ppm or 0.01%), and 0.001-0.05 percent by mass selected from zirconium, strontium, barium, indium, and aluminum (100 ppm or 0.01% strontium) (see pars. 0010-0013 and Table 2, line 74).

## Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to PATRICIA DAVIS whose telephone number is (571)270-7868. The examiner can normally be reached on 7:30am-5pm EST. Monday-Friday, alternate Fridays off.

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If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Brian Sines can be reached on 571-272-1263. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the

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15 P.D.

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/Brian J. Sines/
Supervisory Patent Examiner, Art Unit 4111